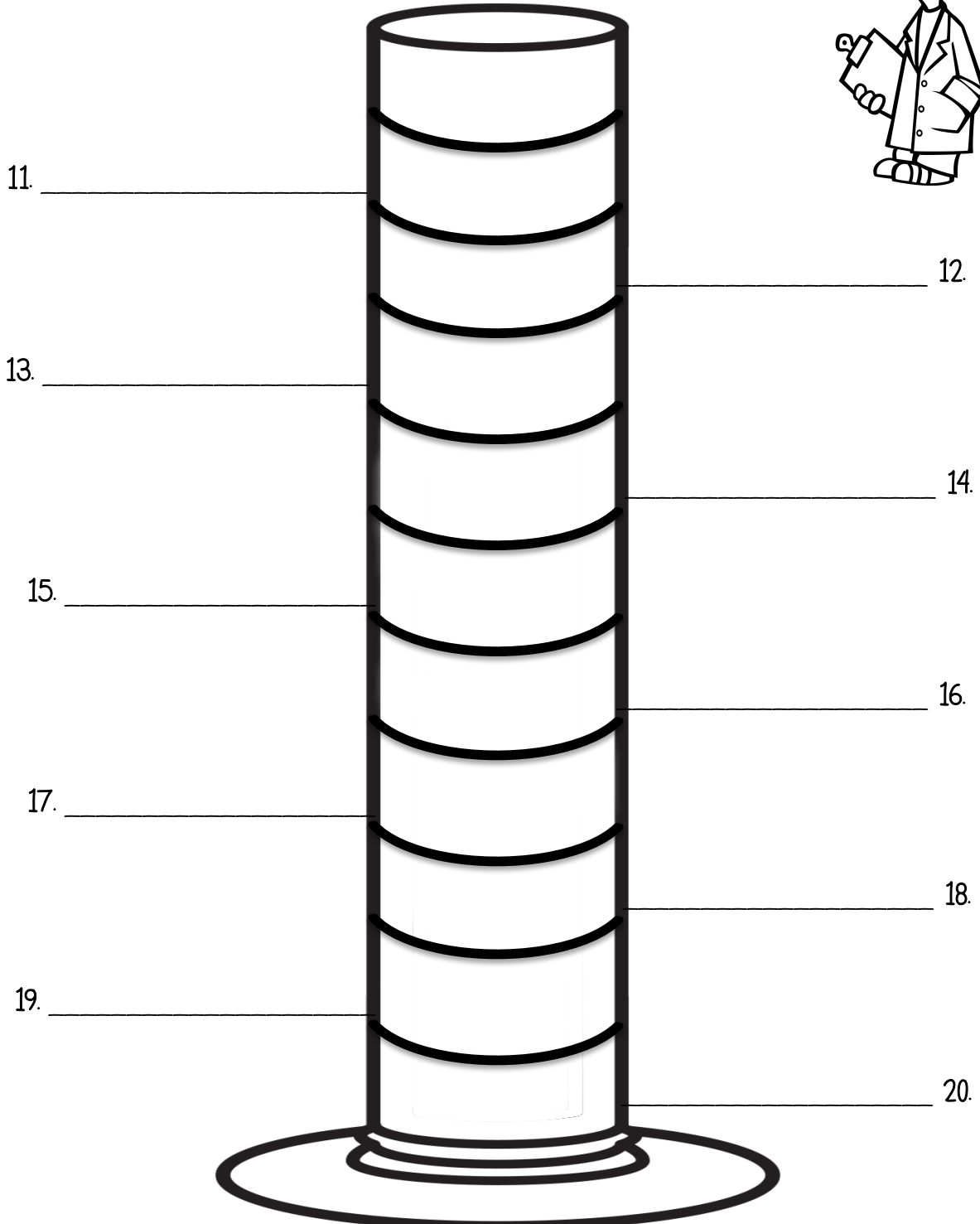


# Exploring Density

**Directions:** A student wants to create a Density Column using some common household materials. First, calculate the density of each substance listed on the back of this paper. Second, identify its position in the column based on its density.



Calculate the density of each substance.  
Make sure to show your work and don't forget to include units!

$$D = \frac{\text{Mass}}{\text{Volume}}$$

1. Water  
M: 68 g V: 68 mL

Density: \_\_\_\_\_

2. Lamp Oil  
M: 32 g V: 40 mL

Density: \_\_\_\_\_

3. Maple Syrup  
M: 8.22 g V: 60 mL

Density: \_\_\_\_\_

4. Rubbing Alcohol  
M: 237 g V: 3.0 mL

Density: \_\_\_\_\_

5. Milk  
M: 5.15 g V: 5.0 mL

Density: \_\_\_\_\_

6. Baby Oil  
M: 332 g V: 4.0 mL

Density: \_\_\_\_\_

7. Dish Soap  
M: 53 g V: 5.0 mL

Density: \_\_\_\_\_

8. Vegetable Oil  
M: 6.44 g V: 7.0 mL

Density: \_\_\_\_\_

9. Honey  
M: 7.1 g V: 5.0 mL

Density: \_\_\_\_\_

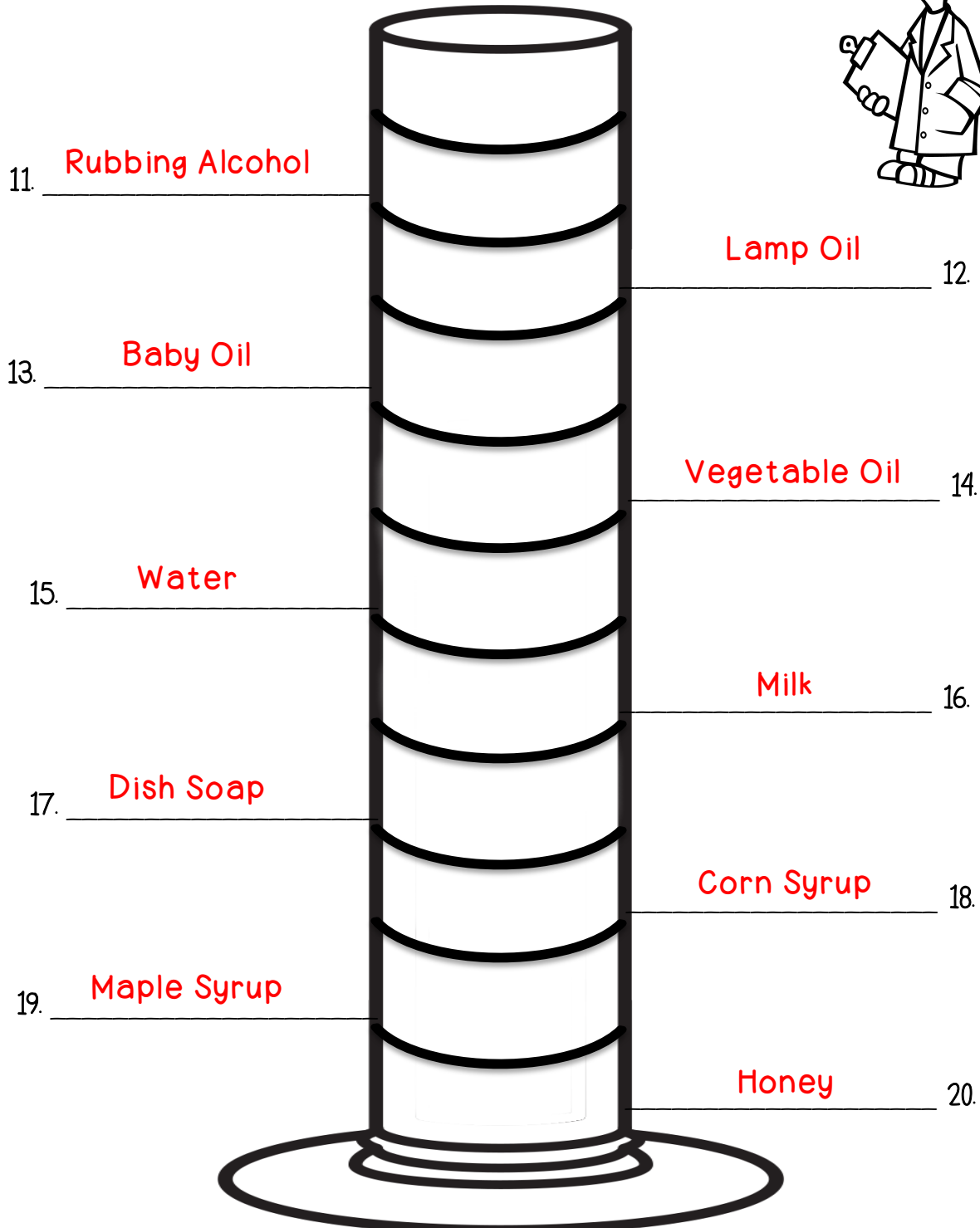
10. Corn Syrup  
M: 7.98 g V: 6.0 mL

Density: \_\_\_\_\_



# Exploring Density

**Directions:** A student wants to create a Density Column using some common household materials. First, calculate the density of each substance listed on the back of this paper. Second, identify its position in the column based on its density.



Calculate the density of each substance.  
Make sure to show your work and don't forget to include units!

$$D = \frac{\text{Mass}}{\text{Volume}}$$

1. Water  
M: 68 g V: 68 mL

Density: 1.0 g/mL

2. Lamp Oil  
M: 32 g V: 40 mL

Density: 0.80 g/mL

3. Maple Syrup  
M: 8.22 g V: 6.0 mL

Density: 1.37 g/mL

4. Rubbing Alcohol  
M: 2.37 g V: 3.0 mL

Density: 0.79 g/mL

5. Milk  
M: 5.15 g V: 5.0 mL

Density: 1.03 g/mL

6. Baby Oil  
M: 3.32 g V: 4.0 mL

Density: 0.83 g/mL

7. Dish Soap  
M: 5.3 g V: 5.0 mL

Density: 1.06 g/mL

8. Vegetable Oil  
M: 6.44 g V: 7.0 mL

Density: 0.92 g/mL

9. Honey  
M: 7.1 g V: 5.0 mL

Density: 1.42 g/mL

10. Corn Syrup  
M: 7.98 g V: 6.0 mL

Density: 1.33 g/mL

